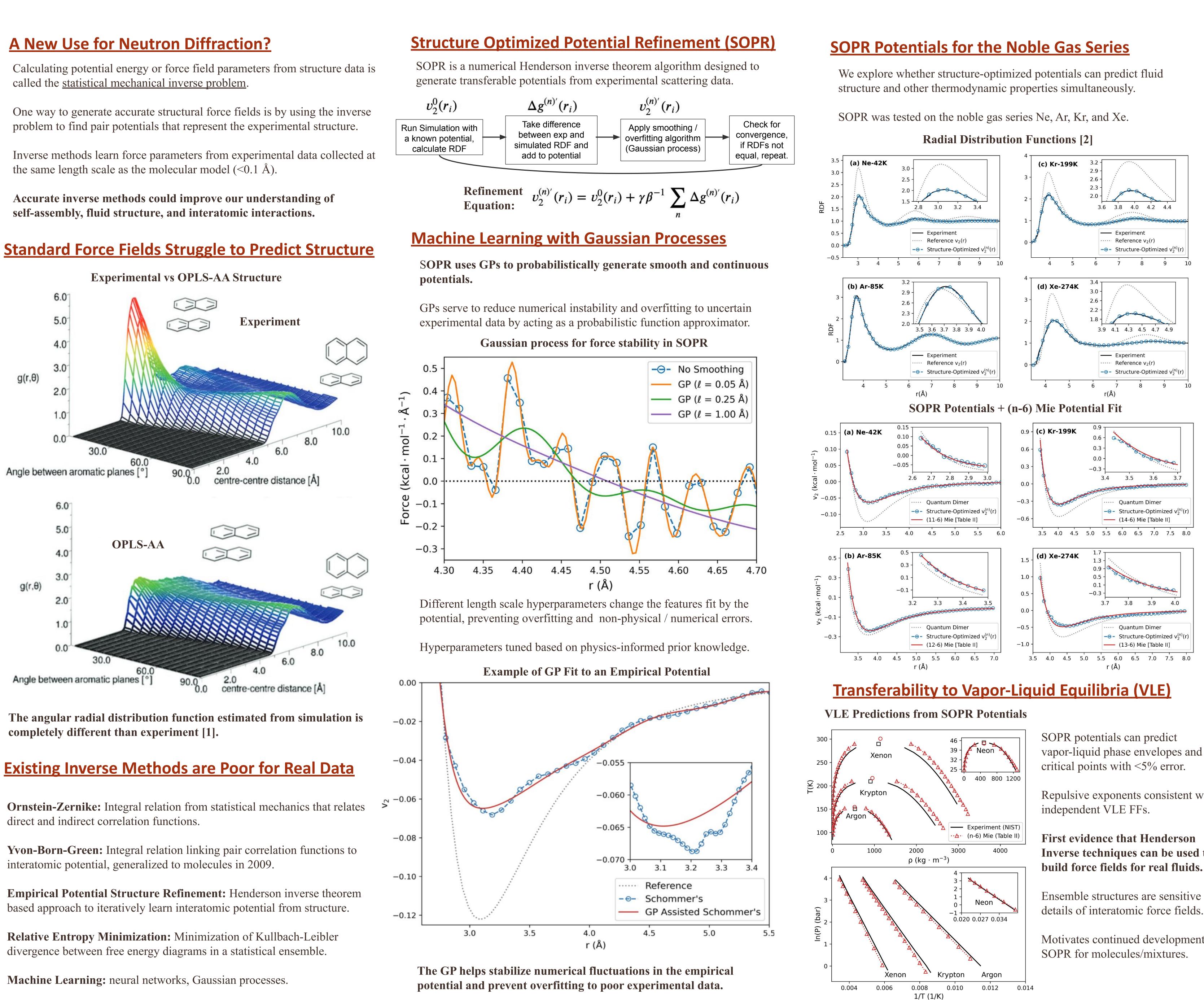
Constructing transferable force fields from neutron scattering measurements with structure optimized potential refinement Brennon L. Shanks, Abdur R. Shazed, Michael P. Hoepfner University of Utah, Department of Chemical Engineering

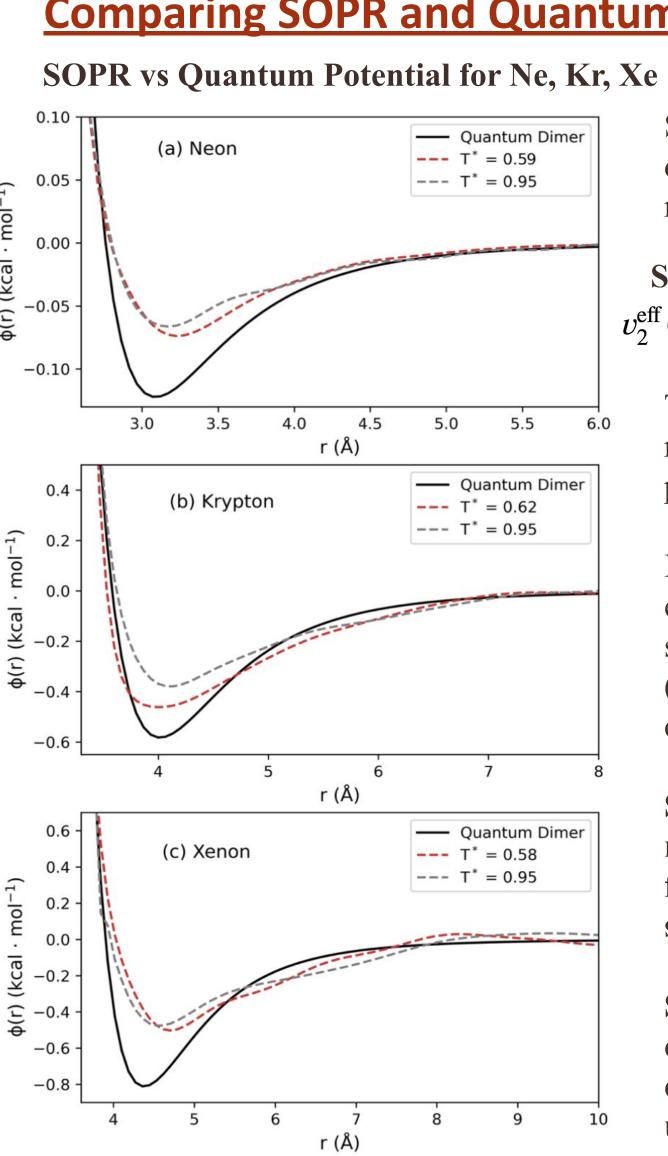


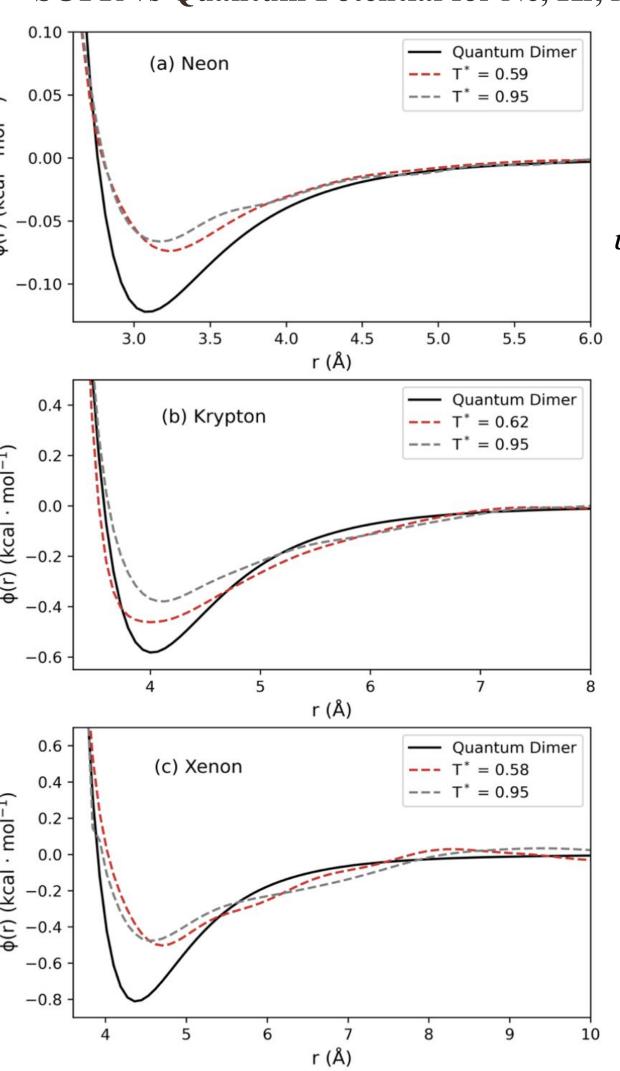
Repulsive exponents consistent with

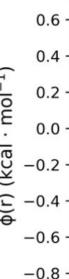
First evidence that Henderson Inverse techniques can be used to build force fields for real fluids.

Ensemble structures are sensitive to details of interatomic force fields.

Motivates continued development of







<u>Remaining Scientific Questions</u>

References

F. Headen, T., L. Cullen, P., Patel, R., Taylor, A. & T. Skipper, N. The structures of liquid pyridine and naphthalene: the effects of heteroatoms and core size on aromatic interactions. Phys. Chem. Chem. Phys. 20, 2704-2715 (2018). Shanks, B. L., Potoff, J. J. & Hoepfner, M. P. Transferable Force Fields from Experimental Scattering Data with Machine Learning Assisted Structure Refinement. J. Phys. Chem. Lett. 11512–11520 (2022) doi:10.1021/acs.jpclett.2c03163.

Acknowledgements

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<u>Comparing SOPR and Quantum Pair Potentials</u>

SOPR potentials can be decomposed into quantum and many-body pair terms.

SOPR Quantum Many-Body $v_2^{\text{eff}}(\mathbf{r}_{ij}; \rho, T) = v_2(\mathbf{r}_{ij}) + v_2^m(\mathbf{r}_{ij}; \rho, T)$

The quantum potential represents a two-body atomic potential in a vacuum.

In noble gases, many-body effects appear to be less significant near the triple point $(T^* = 0.6)$ compared to the critical point $(T^* = 1)$.

SOPR can be used to quantify many-body interactions as a function of thermodynamic state.

SOPR enables us to study the contribution many-body effects in fluid ensembles using neutron diffraction.

General Conclusions

1. Structure can be used to extract accurate force fields in simple liquids.

2. SOPR potentials are consistent with other FFs for vapor liquid equilibria.

3. SOPR can be used to quantify many-body interactions in fluid ensembles.

Can SOPR be extended to molecular liquids / mixtures?

2. How accurately do SOPR potentials model third and higher order effects?

3. Can SOPR be used to predict thermodynamic properties for systems in extreme conditions or outside of its calibration range?

4. What algorithm changes could improve computational stability?

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Contributors:

Brennon L. Shanks: Conceptualization, coding, algorithm development, vriting, poster preparation and presentation

Abdur R. Shazed: Conceptualization, coding, algorithm development

Aichael P. Hoepfner: Funding acquisition, conceptualization, poster reparation and editing